The First Stereoselective Total Synthesis of a Naturally Occurring Bioactive Diarylheptanoid, (3R,6E)-1,7-Bis(4-hydroxyphenyl)hept-6-en-3-ol, through Two Different Approaches¹)

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The stereoselective total synthesis of a naturally occurring bioactive diarylheptanoid, $(3R,6E)$ -1,7bis(4-hydroxyphenyl)hept-6-en-3-ol, has been accomplished starting from 4-hydroxybenzaldehyde through two different approaches involving Wittig olefination, hydrolytic kinetic resolution of a racemic epoxide, and olefin cross-metathesis reaction as the key steps.

Introduction. – Diarylheptanoids exhibit various important biological activities including anticancer, anti-HIV, anti-inflammatory, and antioxidant properties [1]. A large number of these compounds with different functionalities have been isolated from natural sources [2]. Recently, a new diarylheptanoid, $(3R,6E)$ -1,7-bis(4-hydroxyphenyl)hept-6-ene-3-ol (1) and its enantiomer have been isolated in very small quantities from the rhizomes of Curcuma kwangrsiensis (Zingiberaceae) [3]. The compound was evaluated for its inhibitory effect on NO production induced by lipopolysaccharide-activated macrophages. The activity of the compound was found to be close to that of indomethacin [3]. The total synthesis of the compound has not yet been reported. In continuation of our work [4] on the construction of bioactive natural products, we have accomplished the stereoselective total synthesis of 1, which we report here.

Results and Discussion. – The retrosynthetic analysis (*Scheme 1*) revealed that 1 can be synthesized from the alkenes 2 and 3. The first one, 2, can be prepared from the epoxide 4, which in turn can be synthesized from the aldehyde 5 generated from 4 hydroxybenzaldehyde (6). The diarylheptanoid 1 can also be synthesized (Scheme 2) from the aldehyde 7 and the phosphonium ylide 8. The aldehyde 7 can be prepared from the triol 9, which can again be produced from the olefinic alcohol 2 obtained from 6.

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Scheme 1. First Retrosynthetic Analysis of 1

Scheme 2. Second Retrosynthetic Analysis of 1

The present synthesis of 1 was initiated (Scheme 3) by converting 4-hydroxybenzaldehyde (6) to its benzyl ether, which was subjected to Wittig olefination with $PPh_3CHCOOEt$ and subsequent reduction with NaBH₄/NiCl₂ · 6 H₂O, followed by adding $LiAlH₄$ to produce the alcohol 10 [5]. The latter was oxidized with PCC to yield the corresponding aldehyde 5, which again underwent Wittig olefination with PPh_3CH_2 to furnish the olefin 11 [6]. Treatment of 11 with m-CPBA afforded the epoxide 12 [6], which, on hydrolytic kinetic resolution [7] with (S, S) -Jacobsen's catalyst (A) , generated the (S) -configured epoxide 4. The latter was then treated with allylmagnesium bromide in the presence of CuI to give the olefinic alcohol 2. Eventually, the crosscoupling metathesis reaction [8] of 2 with 4-ethenylphenol (3) using *Grubbs'* secondgeneration catalyst, B, resulted in the formation of the heptanoid 13. Finally, deprotection of the benzyl ether of 13 with TiCl₄ gave the target molecule 1.

In a second approach, the chiral olefinic alcohol 2 was treated with $OsO₄$ and 4methylmorpholine N-oxide (NMO) in aqueous acetone, and the resulting triol 9 was treated with $NaIO₄$ to form the aldehyde 7 (Scheme 4). The latter was directly subjected to *Wittig* olefination with the phosphonium ylide $\boldsymbol{8}$ to produce 13 as the major product along with its (Z) -isomer (isomeric ratio 8:2). The product 13 was separated, and subsequently its benzyl ether group was deprotected with TiCl_4 to afford

a) Pyridinium chlorochromate (PCC), Celite, CH₂Cl₂, 0° to r.t, 2 h; 81%. b) BuLi, THF, Ph₃P=CH₂, -78° to r.t, 4 h; 76%. c) meta-Chloroperbenzoic acid (m-CPBA), CH₂Cl₂, NaHCO₃, 0° to r.t, 3 h; 86%. d) $((S, S)$ -Salen)Co(OAc) $({\bf A})$, dist. H₂O, THF, 0°, 24 h; 45%. *e*) C₃H₃MgBr, CuI, THF, -78° , 4 h; 81%. f) Grubbs' second generation catalyst (B), CH₂Cl₂, 50°, 4 h; 68%. g) TiCl₄, CH₂Cl₂, 0°, 3 h; 72%.

compound 1. The optical rotation and spectroscopic data of the latter were found to be identical to those of the naturally occurring $(3R,6E)$ -1,7-bis(4-hydroxyphenyl)hept-6en-3-ol.

Conclusions. – In conclusion, we have developed the first stereoselective total synthesis of the natural bioactive heptanoid, (3R,6E)-1,7-bis(4-hydroxyphenyl)hept-6 en-3-ol starting from commercially available 4-hydroxybenzaldehyde through two different approaches, employing hydrolytic kinetic resolution of a racemic epoxide, Wittig olefination, and olefin cross metathesis reaction as the key steps. Both of the

Scheme 3

a) OsO₄, NMO, aq. acetone, r.t, 4 h. b) NaIO₄, NaHCO₃, CH₂Cl₂, r.t, 2 h. c) BuLi, THF, -78° , 4 h, 67%. d) TiCl₄, CH₂Cl₂, 0°, 3 h; 72%.

synthetic methods are simple and can be utilized for the generation of various analogs of 1.

Experimental Part

General. Silica-gel F_{254} plates were used for TLC; the spots were examined under UV light and then developed by an I₂ vapor. Column chromatography (CC) was performed with silica gel (SiO₂; *BDH* 100 – 200 mesh). Solvents were purified according to standard procedures. Org. extracts were dried over anh. Na₂SO₄. Optical rotations: *JASCO DIP 300* digital polarimeter, at 25°. IR Spectra: *Perkin-Elmer* RX FT-IR spectrophotometer. NMR Spectra: Varian Gemini 200- (1H) and 50-MHz (^{13}C) spectrometer. ESI-MS: VG-Autospec micromass.

 $1-(Benzyloxy)-4-(but-3-en-1-yl)benzene$ (11) [6]. To a stirred suspension of *Celite* (10.0 g) in dry CH_2Cl_2 (30 ml), a soln. of 3-[4-(benzyloxy)phenyl]propan-1-ol (10; 12.5 g, 51.6 mmol) in dry CH_2Cl_2 (10 ml) at r.t. was added, followed by careful addition of PCC (22.1 g, 103.3 mmol). The mixture was allowed to stir slowly at r.t. After stirring for 1 h, the mixture was filtered through sintered funnel with CH_2Cl_2 (2 \times 30 ml) The combined org. extracts were washed with brine, dried, and concentrated under reduced pressure. The residue was purified by CC (hexane/AcOEt) to afford 3-[4-(benzyloxy)phenylpropanal (5; 10.41 g, 81%) as a colorless liquid.

To a stirred soln. of methylidene(triphenyl)phospine (50.05 g, 125.0 mmol) in dry THF (30 ml) under N_2 at -78° , BuLi (41.6 ml in 2.5m hexane) was added dropwise, and the mixture was stirred at that temp. for a few min until the yellow color appeared. Then, 5 (10.0 g, 41.66 mmol) in THF (20 ml) was added dropwise, and the mixture was stirred at that temp. for 3 h. The reaction was quenched with sat. NH₄Cl soln. (20 ml) at 0° , and the mixture was extracted with AcOEt (2×50 ml). The combined org. extracts were washed with brine, dried, and concentrated under reduced pressure. The residue was purified by CC (hexane/AcOEt) to afford 11 (7.53 g, 76%). Pale yellow liquid. IR (neat): 2925, 2855, 1460, 1376, 1250. ¹H-NMR (200 MHz, CDCl₃): 7.20 – 7.11 $(m, 5H)$; 7.02 $(d, J = 8.0, 2H)$; 6.82 $(d, J = 8.0, 3H)$ 2 H); 5.90 – 5.71 (m, 1 H); 5.02 – 4.95 (m, 2 H); 4.99 (s, 2 H); 2.62 (t, $J = 7.0$, 2 H); 2.39 – 2.22 (m, 2 H). ¹³C-NMR (50 MHz, CDCl₃): 157.8; 139.1; 138.6; 134.9; 130.0; 129.1; 127.5; 126.9; 115.2; 115.0; 70.0; 32.5; 31.4. ESI-MS: 239 ($[M + H]^+$). Anal. calc. for C₁₇H₁₈O (238.32): C 85.67, H 7.61; found: C 85.77, H 7.63.

2-{2-[4-(Benzyloxy)phenyl]ethyl}oxirane (12) [6]. To a stirred soln. of 11 (7.0 g, 29.4 mmol) in dry CH_2Cl_2 (10 ml) under N₂ at 0° was added m-CPBA (6.07 g, 35.2 mmol) dissolved in dry CH₂Cl₂ (10 ml). The mixture was stirred for 2 h. The reaction was quenched with sat. aq. NaHCO₃ (6 ml), and the mixture was stirred further for 1 h and then extracted with CH₂Cl₂ (2×30 ml). The combined org. layers were dried and concentrated *in vacuo*. The residue on purification by CC (hexane/AcOEt) afforded pure 12 (6.91 g, 86%). Pale yellow liquid.

 $(2S)-2-[2-[4-(Benzyloxy)phenyl]ethyl/oxiran$ e (4). A soln. of 12 (6.0 g, 23.1 mmol) and ((S,S)-Salen)Co(OAc) (A; 0.076 g, 0.12 mmol) in THF (30 ml) was stirred at 0 \degree for 5 min, and then dist. H₂O (229 ml, 12.7 mmol) was added. After stirring for 24 h, this mixture was concentrated and purified by CC (hexane/AcOEt) to afford 4 (2.3 g, 45%). Yellow liquid. $\lbrack \alpha \rbrack_0^{25} = -4.6$ ($c = 1.0$, CHCl₃). IR (neat): 3454, 2926, 2854, 1730, 1253. ¹H-NMR (200 MHz, CDCl₃): 7.42 – 7.12 (m, 5 H); 7.08 (d, J = 8.0, 2 H); 6.82 (d, $J = 8.0, 2$ H); 5.02 (s, 2 H); $2.95 - 2.86$ (m, 1 H); $2.80 - 2.60$ (m, 2 H); $2.41 - 2.32$ (m, 1 H); $2.29 - 2.20$ (m, 1 H); 1.84 – 1.72 (m, 1 H); 1.60 – 1.56 (m, 1 H). 13C-NMR (50 MHz, CDCl3): 155.4; 137.5; 134.1; 129.8; 129.2; 127.6; 126.9; 115.2; 70.2; 51.3; 46.1; 34.8; 29.9. ESI-MS: 255 ($[M + H]^+$). Anal. calc. for C₁₇H₁₈O₂ (254.32): C 80.28, H 7.13; found: C 80.39, H 7.11.

(3R)-1-[4-(Benzyloxy)phenyl]hept-6-en-3-ol (2). A round-bottomed flask was charged with CuI $(0.28 \text{ g}, 1.4 \text{ mmol})$, gently heated under vacuum, and then slowly cooled under a flow of N₂. THF (10 ml) was then added, and the resulting suspension was cooled to -78° , stirred, and allylmagnesium bromide $(9.4 \text{ ml in } 1 \text{M}$ THF) was added. A soln. of 4 (2.0 g, 7.8 mmol) in THF (10 ml) was added to the above reagent, and the mixture was stirred at -78° for 4 h. After consumption of starting material, the reaction was quenched with sat. aq. NH₄Cl. The H₂O layer was extracted with AcOEt (50 ml), and the combined org. layers were washed with brine, dried, and concentrated in vacuo. The residue on purification by CC (hexane/AcOEt) afforded 2 (1.88 g, 81%). Yellow liquid. $\lbrack \alpha \rbrack_0^{25} = +10.3$ ($c = 1.0$, CHCl₃). IR (neat): 3451, 2925, 2855, 1637, 1459. ¹H-NMR (200 MHz, CDCl₃): 7.41 – 7.20 (*m*, 5 H); 7.04 (*d*, *J* = 8.0, 2 H); 6.82 (*d*, $J = 8.0, 2$ H); 5.93 – 5.78 (m, 1 H); 5.38 – 4.93 (m, 1 H); 5.00 (s, 2 H); 3.69 – 3.56 (m, 1 H); 2.75 – 2.52 (m, 2 H); 2.20 – 2.05 (m, 2 H); 1.80 – 1.61 (m, 4 H). ¹³C-NMR (50 MHz, CDCl₃): 156.8; 138.7; 137.5; 134.8; 129.6; 128.5; 127.0; 126.6; 115.0; 114.8; 70.4; 69.8; 39.8; 36.2; 30.8; 30.1. ESI-MS: 297 ($[M + H]$ ⁺). Anal. calc. for $C_{20}H_{24}O_2$ (296.40): C 81.04, H 8.16; found: C 81.16, H 8.09.

 $4-(IESR)-7-I4-(Benzvloxv)phenyl-5-hydroxvhept-I-en-I-vliphenol (13)$. A soln. of 2 (0.5 g, 1.68 mmol) and 3 (0.506 g, 4.22 mmol) in dry CH₂Cl₂ (50 ml) was first bubbled with N₂ flow, after which Grubbs' second-generation catalyst (0.045 g, 0.054 mmol) was added at once, and the resulting mixture was heated under N₂ at 50 $^{\circ}$ for 4 h. After cooling, the solvent was evaporated in vacuo. The residue on purification by CC (hexane/AcOEt) afforded **13** (0.348 g, 68%). Yellow liquid. $[a]_D^{25} = +2.9$ $(c = 1.0, CHCl₃)$. IR (near): 3427, 2926, 2856, 1250. ¹H-NMR (200 MHz, CDCl₃): 7.39 – 7.21 (*m*, 5 H); 7.20 $(d, J = 8.0, 2 \text{ H})$; 7.01 $(d, J = 8.0, 2 \text{ H})$; 6.82 $(d, J = 8.0, 2 \text{ H})$; 6.68 $(d, J = 8.0, 2 \text{ H})$; 6.56 $(d, J = 16.0, 16.0)$ 1 H); 5.81 – 5.69 (m, 1 H); 4.99 (s, 2 H); 3.68 – 3.51 (m, 1 H); 2.58 (t, $J = 7.0$, 2 H); 2.30 – 2.02 (m, 4 H); 1.78 – 1.58 (m, 2 H); 1.48 – 1.30 (m, 2 H). ESI-MS: 389 ([$M + H$]⁺). Anal. calc. for C₂₆H₂₈O₃ (388.50): C 80.38, H 7.26; found: C 80.47, H 7.18.

 $4,4'-[(1E,5R)-5-Hydroxyhept-1-ene-1,7-diy] diphenol (1)$. To a soln. of 13 (0.30 g, mmol) in dry CH_2Cl_2 (5 ml) was added a soln. of TiCl₄ (0.16 ml, 2.64 mmol) in dry CH₂Cl₂ (5 ml) under N₂ at 0°. The reaction was monitored by TLC, until the starting material was consumed. H₂O was added, and the mixture was extracted with CH₂Cl₂. The combined org. extract was washed with brine and dried. The solvent was removed under reduced pressure, and the crude mass was purified by CC (silica gel; hexane/ AcOEt) to afford pure 1 (0.22 g, 74%). Yellow liquid. $\lbrack a \rbrack_0^2 = +15.5$ ($c = 3.0$, CHCl₃). IR (neat): 3423, 1608, 1510, 1453, 1233. ¹H-NMR (200 MHz, CDCl₃): 7.22 $(d, J = 8.0, 2 H)$; 7.02 $(d, J = 8.0, 2 H)$; 6.81 $(d,$ $J = 8.0, 2$ H); 6.68 (d, $J = 8.0, 2$ H); 6.58 (d, $J = 16.0, 1$ H; 5.83 – 5.65 (m, 1 H); 3.64 – 3.53 (m, 1 H); 2.81 – 2.77 (m, 1 H); 2.65 – 2.45 (m, 1 H); 2.22 – 2.01 (m, 2 H); 1.79 – 1.61 (m, 2 H); 1.59 – 1.48 (m, 2 H). 13C-NMR (50 MHz, CDCl₃): 159.2; 156.7; 135.8; 134.6; 129.5; 128.8; 127.5; 126.4; 114.9; 114.1; 70.0; 39.6; 39.3; 30.8; 30.2. ESI-MS: 299 ($[M + H]^+$). Anal. calc. for C₁₉H₂₂O₃ (298.38): C 76.48, H 7.43; found: C 76.58, H 7.34.

 $(5S)$ -7-[4-(Benzyloxy)phenyl]heptane-1,2,5-triol (9). To stirred soln. of 2 (1.0 g, 3.37 mmol) in aq. acetone (10 ml) was added NMO (1.2 g, 10.13 mmol), followed by $OsO₄$ (2.5 weight-% in ⁱBuOH). The mixture was stirred for 3 h. After completion of reaction, the reaction was quenched with sat. NaHSO₃ (5 ml), and the mixture was stirred for 30 min and then extracted with AcOEt (3×20 ml). The combined org. layers were washed with brine, dried, and concentrated in vacuo. The residue, 9, was directly used for the next step.

 $(4S)$ -6-[4-(Benzyloxy)phenyl]-4-hydroxyhexanal (7). To a stirred soln. of 9 (0.9 g, 2.72 mmol) in dry CH₂Cl₂ (10 ml) at 0° , ag. NaHCO₃ (5 ml) was added, followed by careful addition of NaIO₄ (1.1 g, 5.45 mmol). The mixture was allowed slowly to warm to r.t. After stirring for 2 h, Na₂SO₄ (1.0 g) was added, and the mixture was stirred vigorously for 30 min. The mixture was filtered through a sintered funnel with CH₂Cl₂ (2×30 ml). The combined org. extracts were washed with brine, dried, and concentrated under reduced pressure. The residue was purified by CC (hexane/AcOEt) to afford 7 as Yellow liquid.

To a stirred soln. of $[(4-hydroxyphenyl)methylidenel triphenylphosphine (8; 2.9 g, 6.03 mmol) in dry)$ THF (10 ml) under N_2 at -78° BuLi (2.0 ml in 2.5m hexane) was added dropwise and the mixture was stirred at -78° for few min, until a yellow color appeared. Then, 7 was added dropwise, and the mixture was stirred at -78° for 4 h. The reaction was quenched with sat. NH₄Cl soln. (10 ml) at 0°, and the mixture was extracted with AcOEt $(2 \times 50 \text{ ml})$. The combined org. extracts were washed with brine, dried, and concentrated under reduced pressure. The residue was purified by CC(hexane/AcOEt) to afford 13 (0.673 σ , 67%). Yellow liquid.

Compound 13 was deprotected as reported earlier to obtain 1. $\left[\alpha\right]_{\text{D}}^{25} = +15.5$ (c = 3.0, CHCl₃). The optical and spectral properties of the compounds 1 were identical to those of the naturally occurring bioactive diarylheptanoid, 4,4'-[(1E,5R)-5-hydroxyhept-1-ene-1,7-divl]diphenol [7].

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